

## Reactions of ethyl 3-(diethoxyphosphoryl)-3,3-difluoropyruvate with some nucleophilic reagents

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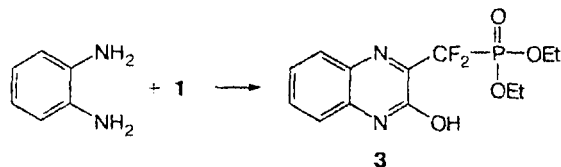
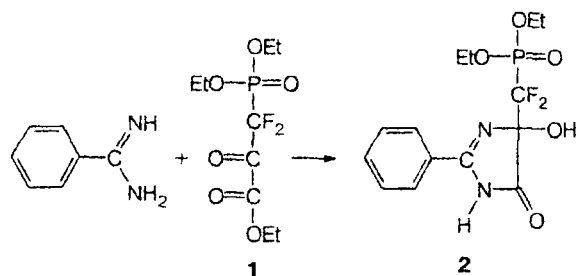
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Reactions of ethyl 3-(diethoxyphosphoryl)-3,3-difluoro-2-oxopropionate with a number of nucleophilic reagents were studied. New procedures were developed for the synthesis of difluoromethylphosphonate-substituted nitrogen heterocycles. The ketoester under study is much less reactive in C-hydroxyalkylation of aromatic amines than methyl 3,3,3-trifluoropyruvate.

**Key words:** ethyl 3-(diethoxyphosphoryl)-3,3-difluoro-2-oxopropionate, heterocyclization, C-hydroxyalkylation, N-hydroxyalkylation, difluoromethylphosphonates, imidazolines, indoles, indolines, quinoxalines, phenylhydrazones.

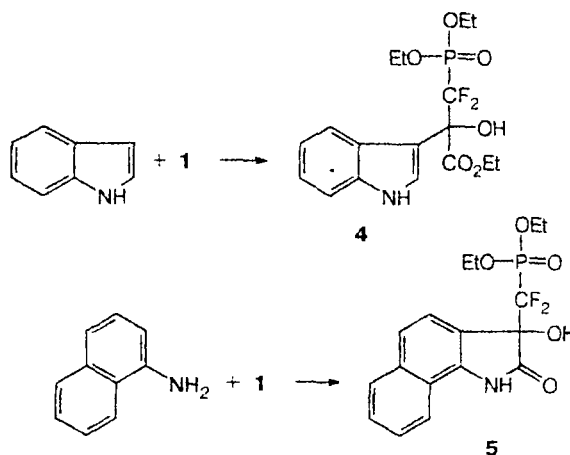
It is known that methyl 3,3,3-trifluoropyruvate vigorously reacts with amidines,<sup>1</sup> *o*-phenylenediamines,<sup>2</sup> and arylamines,<sup>3</sup> is involved into C-hydroxyalkylation of indoles,<sup>4</sup> and reacts with phenylhydrazines to form phenylhydrazones.<sup>5</sup> Great interest in compounds containing the difluoromethylphosphonate group<sup>6</sup> gave impetus to our studies of the reactivity of ethyl 3-(diethoxyphosphoryl)-3,3-difluoropyruvate (**1**),<sup>7</sup> whose chemical properties are virtually unknown. In this work, we studied the reactions of compound **1** with benzamidine, indole, *o*-phenylenediamine, 1-naphthylamine, and phenylhydrazine.

Ketoester **1**, like methyl trifluoropyruvate, exothermically reacts with benzamidine and *o*-phenylenediamine at 20 °C to form the corresponding imidazole and quinoxaline derivatives **2** and **3** in high yields.

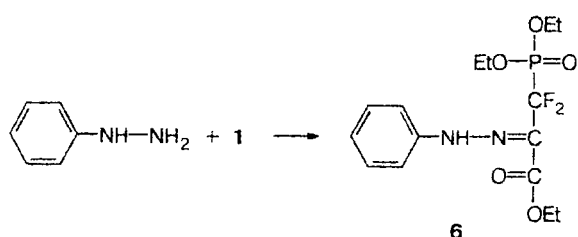


Like methyl trifluoropyruvate, ketoester **1** reacts with indole and (more slowly) with 1-naphthylamine to give

a product of C-hydroxyalkylation of indole **4** and substituted indolinone **5**, respectively. The reactions with indole at room temperature are completed in 20 min, whereas the reaction with 1-naphthylamine under the same conditions is completed in 48 h, which is indicative of a substantial decrease in the reactivity of **1** in reactions with arylamines compared to methyl trifluoropyruvate,<sup>3</sup> whose reaction with 1-naphthylamine is completed in 1 h. Moreover, under these conditions ester **1**, unlike methyl trifluoropyruvate,<sup>3</sup> is not involved in C-hydroxyalkylation of dialkylanilines at the *para* position. Apparently, this is attributable to the higher stability of hemiaminal, which forms in the case of ketoester **1** as a result of competitive reversible N-hydroxyalkylation of the amino group of the substrate.<sup>8</sup>



When heated, ester **1**, like methyl trifluoropyruvate,<sup>5</sup> gives phenylhydrazone (**6**) as a viscous oil.



Therefore, ketoester **1** can be used as a precursor in the synthesis of imidazolones, phenylhydrazones, quinoxalines, indoles, and oxindolines containing the difluoromethylphosphonate group. It was also found that ester **1** is much less active in C-hydroxyalkylation of arylamines than methyl trifluoropyruvate, which does not contain the phosphoryl group.

The structures of the resulting compounds were confirmed by the data of elemental analysis and  $^1\text{H}$ ,  $^{19}\text{F}$ , and  $^{31}\text{P}$  NMR spectroscopy.

### Experimental

The  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra were measured on a Bruker AMX-400 instrument operating at 400.13 and 160.62 MHz, respectively. The  $^{19}\text{F}$  NMR spectra were recorded on a Bruker WP-200 instrument operating at 188.31 MHz. The chemical shifts in the  $^1\text{H}$ ,  $^{19}\text{F}$ , and  $^{31}\text{P}$  NMR spectra were measured relative to  $\text{Me}_4\text{Si}$  (internal standard),  $\text{CF}_3\text{COOH}$  (external standard), and 85%  $\text{H}_3\text{PO}_4$  (external standard), respectively. Ethyl 3-(diethoxyphosphoryl)-3,3-difluoro-2-oxopropionate (**1**) was prepared according to a procedure reported previously.<sup>7</sup>

**4-(Diethoxyphosphoryldifluoromethyl)-4-hydroxy-2-phenyl-2-imidazolin-5-one (2).** Compound **1** (240 mg, 0.83 mmol) was added with stirring to a solution of benzamidine (100 mg, 0.83 mmol) in  $\text{MeCN}$  (5 mL) at 20 °C. After 1 h, the precipitate that formed was filtered off, washed with hexane, and dried on a filter. Imidazolinone **2** was obtained in a yield of 290 mg (80%) as white crystals, m.p. 166–168 °C. Found (%): C, 46.17; H, 4.38; N, 7.87.  $\text{C}_{14}\text{H}_{11}\text{F}_2\text{N}_3\text{O}_5\text{P}$ . Calculated (%): C, 46.41; H, 4.73; N, 7.73.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : 1.22 (m, 6 H, 2 Me); 4.13 (m, 4 H, 2  $\text{CH}_2$ ); 7.35 (br.s, 1 H, OH); 7.60–8.00 (m, 5 H, Ph); 11.70 (br.s, 1 H, NH).  $^{19}\text{F}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : -39.4 (d,  $J_{\text{P-F}} = 99.2$  Hz).  $^{31}\text{P}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : 7.6 (t,  $J_{\text{P-F}} = 99.2$  Hz).

**3-(Diethoxyphosphoryldifluoromethyl)-2-hydroxyquinoxaline (3).** Compound **1** (280 mg, 0.93 mmol) was added with stirring to a solution of *o*-phenylenediamine (100 mg, 0.93 mmol) in  $\text{CHCl}_3$  (10 mL) at 20 °C. After 30 min, the solvent was distilled off *in vacuo*. The residue was crystallized from hexane and dried on a filter. Compound **3** was obtained in a yield of 320 mg (99%) as yellowish crystals, m.p. 141–142 °C. Found (%): C, 46.64; H, 4.22; N, 8.07.  $\text{C}_{13}\text{H}_{15}\text{F}_2\text{N}_3\text{O}_4\text{P}$ . Calculated (%): C, 46.99; H, 4.55; N, 8.43.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : 1.33 (t, 6 H, 2 Me,  $J = 7.4$  Hz); 4.32 (m, 4 H, 2  $\text{CH}_2$ ); 7.40 (m, 2 H, Ar); 7.70 (t, 1 H, Ar,  $J = 7.6$  Hz); 7.83 (d, 1 H, Ar,  $J = 8.1$  Hz); 12.80 (br.s, 1 H, OH).  $^{19}\text{F}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : -29.3 (d,  $J_{\text{P-F}} = 99.2$  Hz).  $^{31}\text{P}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : 8.0 (t,  $J_{\text{P-F}} = 99.2$  Hz).

**3-[1-Carbethoxy-(2-diethoxyphosphoryl)-2,2-difluoroethyl]-1-hydroxyindole (4).** Compound **1** (290 mg, 1 mmol) was added with stirring to a solution of indole (120 mg, 1 mmol) in  $\text{CHCl}_3$  (10 mL) at 20 °C. After 20 min, the reaction mixture was filtered and the precipitate was washed with hexane and dried on a filter. Compound **4** was obtained in a yield of 240 mg (59%) as white crystals, m.p. 162 °C. Found (%): C, 50.23;

H, 5.33; N, 3.48.  $\text{C}_{17}\text{H}_{22}\text{F}_2\text{N}_2\text{O}_6\text{P}$ . Calculated (%): C, 50.37; H, 5.47; N, 3.46.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : 1.25 (m, 9 H, 3 Me); 4.19 (m, 6 H, 3  $\text{CH}_2$ ); 6.60 (br.s, 1 H, OH); 6.90, 7.10 (both t, 1 H each, 2 H(Ar),  $J = 7.5$  Hz); 7.40 (m, 2 H, H(2), H(Ar)); 7.80 (d, 1 H, H(Ar),  $J = 8.1$  Hz); 11.10 (s, 1 H, NH).  $^{19}\text{F}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : -31.2 (center of the AB system,  $J_{\text{P-F}} = 99$  Hz,  $J_{\text{FF}} = 299$  Hz).  $^{31}\text{P}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : 9.9 (t,  $J_{\text{P-F}} = 99$  Hz).

**3-(Diethoxyphosphoryldifluoromethyl)-3-hydroxy-2,3-dihydrobenzo[g]indol-2-one (5).** Compound **1** (200 mg, 0.7 mmol) was added with stirring to a solution of 1-naphthylamine (100 mg, 0.7 mmol) in  $\text{CCl}_4$  (5 mL) at 20 °C. After 48 h, the precipitate that formed was filtered off, washed with hexane, and dried on a filter. Compound **5** was obtained in a yield of 150 mg (56%) as gray crystals, m.p. 164–165 °C. Found (%): C, 52.70; H, 4.54; N, 3.60.  $\text{C}_{17}\text{H}_{18}\text{F}_2\text{N}_2\text{O}_5\text{P}$ . Calculated (%): C, 52.99; H, 4.71; N, 3.64.  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : 0.98 (t, 3 H, Me,  $J = 7.15$  Hz); 1.07 (t, 3 H, Me,  $J = 6.98$  Hz); 3.67–3.83 (m, 2 H,  $\text{CH}_2$ ); 3.98 (m, 2 H,  $\text{CH}_2$ ); 7.18 (s, 1 H, OH); 7.50–8.08 (m, 6 H, Ar); 11.30 (s, 1 H, NH).  $^{19}\text{F}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : -38 (d,  $J_{\text{P-F}} = 99$  Hz).  $^{31}\text{P}$  NMR ( $\text{DMSO}-d_6$ ),  $\delta$ : 7.2 (t,  $J_{\text{P-F}} = 99$  Hz).

**Ethyl 3-diethoxyphosphoryl-3,3-difluoro-2-(phenylhydrazono)propionate (6).** Compound **1** (1 g, 3.5 mmol) was added to a solution of freshly distilled phenylhydrazine (370 mg, 3.5 mmol) in benzene (20 mL). The reaction mixture was azeotroped for 2 h. Then the benzene was distilled off *in vacuo*. The residue was dried *in vacuo* (1 Torr) at 50 °C for 15 min. An orange-red oil was obtained in a yield of 1.36 g (100%). Found (%): C, 47.35; H, 5.88; N, 7.28.  $\text{C}_{15}\text{H}_{21}\text{F}_2\text{N}_2\text{O}_5\text{P}$ . Calculated (%): C, 47.62; H, 5.56; N, 7.41.  $^1\text{H}$  NMR ( $\text{benzene}-d_6$ ),  $\delta$ : 1.08 (t, 3 H, Me( $\text{CO}_2\text{Et}$ ),  $J = 7.14$  Hz); 1.14 (t, 6 H, 2 Me( $\text{P}-\text{OEt}$ ),  $J = 6.98$  Hz); 4.04 (q, 2 H,  $\text{CH}_2$  ( $\text{CO}_2\text{Et}$ ),  $J = 7.14$  Hz); 4.27 (m, 4 H, 2  $\text{CH}_2$ ,  $\text{P}-\text{OEt}$ ); 6.93–7.49 (m, 5 H, Ph); 12.68 (s, 1 H, NH).  $^{19}\text{F}$  NMR ( $\text{benzene}-d_6$ ),  $\delta$ : -27.3 (d,  $J_{\text{P-F}} = 102$  Hz).  $^{31}\text{P}$  NMR ( $\text{benzene}-d_6$ ),  $\delta$ : 4.0 (t,  $J_{\text{P-F}} = 102$  Hz).

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